

Some properties of Thomson's atom

In Thomson's model for the Hydrogen atom, the positive charge e is uniformly distributed within a sphere of radius a_0 . The electron, having charge $-e$, is considered to be a point particle and is located inside the sphere.

- a) Find the electric field and the potential generated by the positive charge and the equilibrium position for the electron (assumed to have zero angular momentum).
- b) Find the ionization energy U_I (i.e. the energy needed to remove the electron from the atom) and the related value of a_0 consistent with the experimental value $U_I = 13.6 \text{ eV} = 2.18 \times 10^{-18} \text{ J}$.
- c) Find the oscillation frequency of the electron around its equilibrium position.
- d) Find the polarizability α of the atom and the dielectric constant ϵ in the "solid state" (in which all atoms are adjacent to each other forming a lattice), using the mean field approximation.

Solution

a) The field is radial and can be found by Gauss's theorem applied to an uniformly charged sphere of density $\rho_+ = e/(4\pi a_0^3/3)$:

$$E(r) = \frac{e}{4\pi\epsilon_0} \times \begin{cases} r/a_0^3 & (r < a_0) \\ 1/r^2 & (r > a_0) \end{cases} .$$

The electric potential $V = V(r)$ satisfies $E = -\partial_r V$ and thus we obtain by integrating

$$V(r) = \frac{e}{4\pi\epsilon_0} \times \begin{cases} (3/2a_0 - r^2/2a_0^3) & (r < a_0) \\ 1/r & (r > a_0) \end{cases} ,$$

where we assumed $V(r = \infty) = 0$. The position of stable equilibrium is at $r = 0$.

b) Assuming that the electron is at $r = 0$, its total energy is $-eV(0)$. The energy needed to take the electron to infinity is

$$U_I = eV(0) = \frac{3k_0e^2}{2a_0} ,$$

where $k_0 = 1/4\pi\epsilon_0 \simeq 9 \times 10^9$ SI units. We thus obtain

$$a_0 = \frac{3k_0e^2}{2U_I} = \frac{3 \times 9 \times 10^9 \times (1.6 \times 10^{-19})^2}{2 \times 10^{-18}} \text{ m} = 1.6 \times 10^{-10} \text{ m} .$$

The actual atomic dimensions are about one third of this value.

c) The equation of motion of the electron is

$$\frac{dv_r}{dt} = -\frac{e}{m_e} E(r) = -\frac{k_0e^2}{m_e a_0^3} r \equiv -\omega^2 r ,$$

thus the electron motion is harmonic, with a frequency

$$\omega = \sqrt{\frac{k_0e^2}{m_e a_0^3}} \simeq 7.9 \times 10^{15} \text{ s} .$$

This value approaches the experimental one just as an order of magnitude estimate.

d) Let us find the electric dipole \mathbf{p} induced by an external field \mathbf{E}_0 . Due to this latter, the equilibrium position for the electron shifts to the value r_{eq} given by $\mathbf{E}(r_{eq}) = \mathbf{E}_0$, from which we obtain $\mathbf{r}_{eq} = 4\pi\epsilon_0 a_0^3 \mathbf{E}_0 / e$. The induced dipole is thus

$$\mathbf{p} = -e\mathbf{r}_{eq} \equiv \alpha \mathbf{E}_0 , \quad \alpha = 4\pi\epsilon_0 a_0^3 .$$

The number density of atoms in the "solid" state is $n = 1/(2a_0)^3$; thus

$$\epsilon_r = 1 + \frac{n\alpha}{\epsilon_0} = 1 + \frac{\pi}{2} .$$